不溶性アノードによる硫酸銅めっき添加剤消耗原因の検討

萩原 秀樹*,小合 康裕**,石川 久美子**,君塚 亮一***

Study on Accelerated Additive Consumption by the Use of Insoluble Anode in Acid Copper Plating

Hideki HAGIWARA*, Yasuhiro OGO**, Kumiko ISHIKAWA** and Ryoichi KIMIZUKA***

- * 荏原ユージライト株式会社戦略マーケティング部(〒110-0016 東京都台東区台東4-19-9 山口ビル7)
- ** 荏原ユージライト株式会社総合研究所第一開発室(〒215-0033 神奈川県川崎市麻生区栗木2-4-3)
- *** 荏原ユージライト株式会社総合研究所(〒 215-0033 神奈川県川崎市麻生区栗木 2-4-3)
- *Strategic Marketing Department, Ebara-Udylite Co., Ltd. (Yamaguchi Bldg. 7, 4-19-9 Taitoh, Taitoh-ku, Tokyo 110-0016)
- **1st Development Office, Research & Development Center of Chemicals, Ebara-Udylite Co., Ltd. (2-4-3 Kurigi, Asao-ku, Kawasaki-shi, Kanagawa 215-0033)
- ***Research & Development Center of Chemicals, Ebara-Udylite Co., Ltd. (2-4-3 Kurigi, Asao-ku, Kawasaki-shi, Kanagawa 215-0033)

Abstract

It is widely known that the use of insoluble anodes result in a greater quantity of additive consumption than phosphorised copper anodes. Consequently, the insoluble anode is covered with a permeable membrane in use, but recently, insoluble anodes that do not cause considerable additive consumption without permeable membrane have been reported. This paper reports on the result of our study on the cause of substantial additive consumption that results from the use of insoluble anodes. As a result of the study, it is clearly suggested that in the acid copper plating bath containing chloride ions, hypochlorite is generated at the surface of the insoluble anode by the chloride oxidation reaction, and the hypochlorite decompose the additives by oxidation.

Key Words: Insoluble Anode, Acid Copper Plating, Permeable Membrane, Oxidizing Agent, Chloride Ion